

Functional Groups

1. Hydroxyl $-OH$
2. Carbonyl $\begin{matrix} O \\ || \\ -C- \end{matrix}$ $\begin{matrix} O \\ // \\ C-H \end{matrix}$
3. Amino $\begin{matrix} H \\ | \\ N \\ | \\ H \end{matrix}$
4. Carboxyl $\begin{matrix} O \\ || \\ C-OH \end{matrix} \rightarrow \begin{matrix} O \\ || \\ C-O^- \end{matrix} + H^+$

Review of Key Terms

Collect examples as we go along

- **Macromolecule**
- **Monomer**
- **Polymer**
- **Dehydration synthesis**
- **Hydrolysis**
- **Large molecules with subunits**
- **Single subunit of a macromolecule**
- **Many subunits hooked together**
- **To put together and lose water**
- **To break apart by adding water**

Lipids - 1

C, H, O

Hydrophobic

Energy forms:

FAT, OIL
stores **2x** the energy of carbs

Structural:

Phospholipid, wax

1. Fats store large amounts of energy

- Although fats are not strictly polymers, they are large molecules assembled from smaller molecules by **dehydration** reactions.
- A **fat** is constructed from two kinds of smaller molecules, **glycerol** and **fatty acids**.

Figure 1. Structure of Glycerol

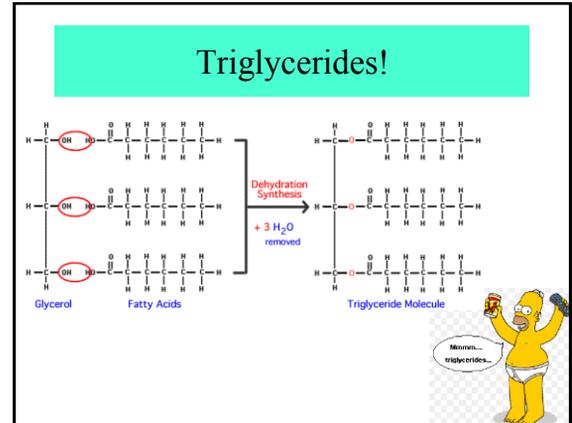
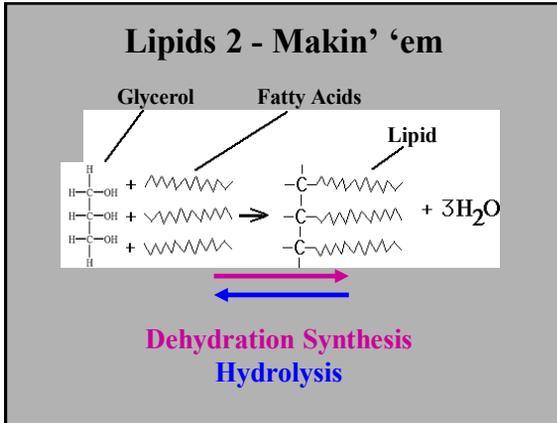
- **Glycerol** consists of a three carbon skeleton with a hydroxyl group attached to each.
- A **fatty acid** consists of a carboxyl group attached to a long hydrocarbon skeleton, often 16 to 18 carbons long.

Fatty acid (Palmitic acid)

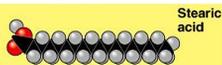
The many nonpolar C-H bonds in the long hydrocarbon skeleton make fats **hydrophobic**.

Glycerol
(a) Dehydration synthesis

Fig. 5.10a



- If there are no carbon-carbon double bonds, then the molecule is a **saturated fatty acid** - a hydrogen at every possible position.



(a) Saturated fat and fatty acid
Fig. 5.11a

- If there are one or more carbon-carbon double bonds, then the molecule is an **unsaturated fatty acid** - formed by the removal of hydrogen atoms from the carbon skeleton.



Double bond causes bending
(b) Unsaturated fat and fatty acid

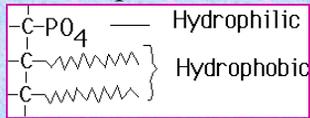
- Fats with saturated fatty acids are **saturated** fats.
 - Most **animal fats** are saturated.
 - Saturated fats are **solid** at room temperature.
 - A diet rich in saturated fats may contribute to cardiovascular disease (**atherosclerosis**) through plaque deposits.
- Fats with unsaturated fatty acids are **unsaturated** fats.
 - Plant and fish fats, known as **oils**, are liquid at room temperature.



- The major function of fats is **energy storage**.
- A gram of fat stores more than twice as much energy as a gram of a polysaccharide.**
- Fat also functions to **cushion** vital organs.
- A layer of fats can also function as **insulation**.
- This subcutaneous layer (**blubber**) is especially thick in whales, seals, and most other marine mammals.

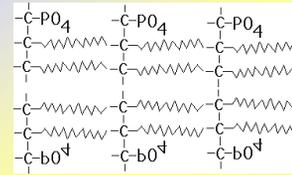


Lipids 3



- This is a phospholipid. Why?
- Phospholipids have both **hydrophilic** and **hydrophobic** regions.
- So, part of the molecule “dissolves” in water and part doesn’t.

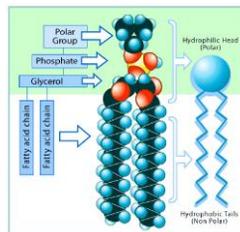
Phospholipids



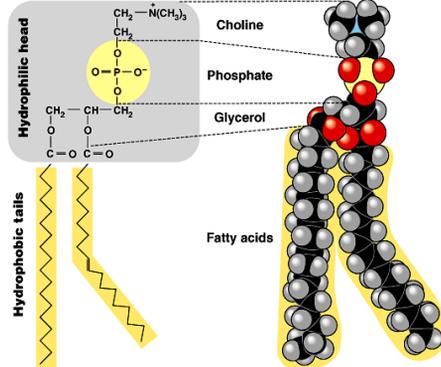
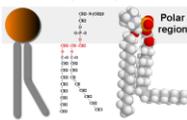
- When phospholipids are placed in water, they spontaneously form a bilayer.
- Because of this, cell membranes have a phospholipid bilayer.

- Phospholipids have **two fatty acids** attached to glycerol and a **phosphate group** at the third position.

- The phosphate group carries a negative charge.
- The fatty acid tails are hydrophobic, but the phosphate group and its attachments form a hydrophilic head.



Phospholipids

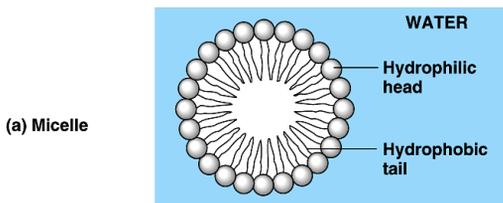


(a) Structural formula

(b) Space-filling model

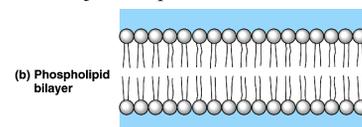
- When phospholipids are added to water, they self-assemble into clumps with the **hydrophobic tails pointing toward the center and the hydrophilic heads on the outside**.

- This type of structure is called a micelle.

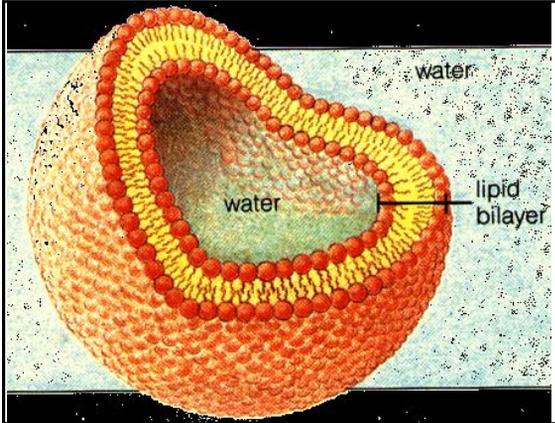


(a) Micelle

- At the surface of a cell phospholipids are arranged as a bilayer.
- Again, the hydrophilic heads are on the outside in contact with the aqueous solution and the hydrophobic tails from the core.
- The phospholipid bilayer forms a barrier between the cell and the external environment.
- They are the major component of membranes.



(b) Phospholipid bilayer



3. Steroids include cholesterol and certain hormones

- Steroids have four carbon rings and a chain of carbons attached.
 - Include hormones, vitamins, and other cellular components
- Cholesterol, an important steroid, is a component in animal cell membranes.
 - Cholesterol is also the precursor from which all other steroids are synthesized.
- Many of these other steroids are hormones, including the vertebrate sex hormones.
- While cholesterol is clearly an essential molecule, high levels of cholesterol in the blood may contribute to cardiovascular disease.

